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Reactive oxygen and nitrogen species (ROS and RNS, respectively) are part of any aerobic lifestyle/metabolism: low (i.e., subtoxic) concentrations of selected ROS and RNS are continuously produced inside and outside the cell by a number of pathways, either accidentally or purposefully. In all eukaryotic cells, the mitochondrial electronic transport chain is the main endogenous source during cellular respiration. Activated phagocytes also release significant amounts of various ROS and RNS by NADPH oxidase, myeloperoxidase, and nitric oxide synthase activity while attacking microorganisms or damaged host cells. Other biologically significant sources of ROS and/or RNS include ionizing radiation, cytochrome p450 activity, the enzymatic system of hypoxanthine/xanthine oxidase, especially in ischemia/reperfusion, metal catalyzed reactions, osmotic stress, and chemotherapeutic drugs (Halliwell & Gutteridge, 2007; Jomova et al., 2010).

Due to their high chemical reactivity, ROS and RNS can modify and oxidize various biological molecules, often altering their biological function, such as unsaturated lipids, carbohydrates, nucleic acids, but mostly, because of their high abundance, proteins. These oxidized (and often damaged) cellular molecules can cause toxicity as such and/or may degrade to form further toxic products, such as reactive carbonyl species (RCS) generated by peroxidation of polyunsaturated fatty acids (PUFAs). For example, highly reactive α,β -unsaturated aldehydes/hydroxyl-alkenals such as 4-hydroxy-2-nonenal and 4-hydroxy-2-hexenal derive from the degradation of peroxidized $n-6$ and $n-3$ PUFAs (Cortese et al., 2009; Fritz & Petersen, 2011; Fritz & Petersen, 2013). RCS can, in turn, react with the nucleophilic sites of proteins, binding to the sulfhydryl group of cysteine and the amino group of lysine or the imidazole group of histidine residues to form Michael or Schiff base protein adducts, known as advanced

oxidation, may thus be an early cellular response to mild oxidative stress and may also play an important role in redox signaling pathways (Dalle-Donne et al., 2007, 2009; D'Autreaux & Toledano, 2007; Brandes, Schmitt, & Jakob, 2009; Rudolph & Freeman, 2009; Zhang et al., 2011; Higdon

As discussed above, there is much evidence that protein thiol oxidation occurs not only as a consequence of oxidative/nitrosative stress conditions, but this modification plays a crucial role in redox signaling pathways in the healthy cell. Amongst the different post-translational oxidative modifications that can occur at the cysteine thiol group, cysteine sulfenic acid (CySOH) is a key player in redox regulation of protein functions under both physiological and oxidative stress conditions and can mediate the transduction of the intracellular signal, hydrogen peroxide, acting as a second messenger, into a biological response (Poole & Nelson, 2008; Haskew-Laytona et al., 2010; Roos & Messens, 2011). However, relatively few molecular details of how this oxidant acts to regulate protein function are currently understood. Furdul and Poole (2013) describe in detail primarily classical and emerging chemical tools and approaches that can be applied to study protein sulfenylation in biological systems, also providing some of the biologically meaningful data that have been collected using such approaches, including demonstration of CySOH formation in IQGAP, a VEGF receptor binding scaffold protein involved in ROS-dependent endothelial cell migration and post-ischemic angiogenesis. Nitrosylation of proteins as well, that is, the addition of an NO group to a Cys thiol to form an S-nitrosoprotein, plays a regulatory role, mediating many of nitric oxide actions and participating in both physiological and pathophysiological processes (Murphy et al., 2012; Piantadosi, 2012; Maron, Tang, & Loscalzo, 2013; Nakamura et al., 2013). In this special issue, Lopez-Sanchez, López-Pedrerera, & Rodríguez-Ariza (2013) exhaustively present up-to-date advances in proteomic methods that are providing researchers with improved tools for exploring protein sulfenylation. In addition, they also review some recent studies of the

products involve some form of carbonylation, even though different structures are associated with carbonyl groups, with the most reactive and common of these being in the form of lipid peroxidation-derived aldehydes. Carbonyl groups may be introduced within the protein primary structure at different sites and

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